

KERALA PUBLIC SCHOOLS
HOME ASSIGNMENT (6TH to 11TH July 2020)



CLASS	SUBJECT	TOPIC / CHAPTER	MODULE / ASSIGNMENT	REFERENC E LINKS
XI	CHEMISTRY	ATOMIC STRUCTURE	Problem 1: Why Bohr's orbits are called stationary states? Problem 2: Explain why the electronic configuration of Cu is $3d^{10}4s^1$ and not $3d^94s^2$. Problem 3: Fe^{3+} ion is more stable than Fe^{2+} ion. Why? Problem 4: Calculate the accelerating potential that must be applied to a proton beam to give it an effective wavelength of 0.005 nm. Problem 5: Give one example of isodiapheres.	

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